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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.002 Å Disorder in solvent or counterion R factor = 0.033 wR factor = 0.081 Data-to-parameter ratio = 19.0

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4-Ammonio-1-hydroxy-2,2,6,6-tetramethylpiperidin-1-ium dichloride monohydrate

The crystal structure of the title compound, $C_9H_{22}N_2O^{2+}\cdot 2Cl^-\cdot H_2O$, has been determined at 173 (1) K. Attached to the piperidinyl ring, which adopts a chair conformation, the hydroxyl and the exocyclic ammonium groups are mutually *trans*, both occupying equatorial positions. The water molecule and the two chloride anions are involved in strong hydrogen bonding *via* $O \cdots H$ and $H \cdots Cl$ interactions.

Comment

The title compound, (I), was first synthesized in an unhydrated form and studied for its protective effect, in Chinese hamster V79 cells, against oxidative damage caused by exposure to H_2O_2 or ionizing radiation (Krishna *et al.*, 1998). Recently, (I) has been found to be a fibre-reactive radical scavenger that binds to lignin-rich wood pulps in aqueous media and inhibits the light-induced oxidation and yellowing of the pulps and papers made from such pulps (Hu, 2003). The molecular structure of this compound is shown in Fig. 1, while selected bond lengths, torsion angles and bond angles are listed in Table 1.



Two different N-H bond lengths and N-C bond lengths are found in the molecule, *viz.* N1-H10 = 0.88 (2) Å, N2-H12 or -H13 or -H14 = 0.98-1.01 (2) Å; and N1-C1 or -C5 = 1.531-1.534 (2) Å and N2-C3 = 1.495 (2) Å. All C-C-C bond angles fall in the range 109.0 (2)-113.0 (2)°, whereas all N1-C-C bond angles fall in the range 105.8 (1)-112.3 (1)°. The piperidinyl ring has a chair conformation, as shown by the torsion angles in Table 1; those involving N1 deviate slightly from those of a perfect chair conformation because of the marginally longer piperidinyl N-C bonds than the C-C bonds. The hydroxyl and exocyclic ammonium groups are mutually *trans*, in equatorial positions. The H₂O molecule is hydrogen bonded *via* intramolecular interactions; the chloride anions are also hydrogen bonded to H atoms on N2 and N1 (see Table 2). Received 18 September 2003 Accepted 22 September 2003 Online 30 September 2003



Figure 1

The molecular structure of (I), with the atom-numbering scheme and 50% probability displacement ellipsoids. H atoms are shown as spheres of an arbitrary radius.

Experimental

Compound (I) was synthesized from 490 mg (2.86 mmol) of 4-amino-2,2,6,6-tetramethylpiperidine-1-oxyl (Aldrich) and 0.74 ml of 11.6 M HCl (8.58 mmol) in 3.0 ml of ethanol according to a literature procedure used for the synthesis of other piperidin-1-ol hydrochloride derivatives (Sosnovsky & Cai, 1995). At the end of the reaction (2 h at room temperature), 20 ml of diethyl ether was added and the mixture cooled in a refrigerator (277 K) overnight. The precipitated white solid (498 mg, 66% yield) was isolated by suction filtration, followed by 4 h of vacuum drying. White crystals were produced by dissolving (I) in 3.0 ml ethanol, adding 20 ml of diethyl ether, and cooling the mixture at 277 K overnight.

Crystal data

$C_{9}H_{22}N_{2}O^{2+}2CI^{-}H_{2}O$ $M_{r} = 263.21$ Orthorhombic, <i>Pbca</i> a = 13.2668 (6) Å b = 9.1840 (5) Å c = 24.471 (1) Å $V = 2981.6 (5) Å^{3}$ Z = 8	Mo K α radiation Cell parameters from 13231 reflections $\theta = 2.2-27.9^{\circ}$ $\mu = 0.42 \text{ mm}^{-1}$ T = 173.2 K Platelet, colourless $0.50 \times 0.30 \times 0.10 \text{ mm}$
$D_x = 1.1/3 \text{ Mg m}^2$ $Data \ collection$ Rigaku/ADSC CCD diffractometer ω and φ scans Absorption correction: multi-scan (Blessing, 1995)	2001 reflections with $I > 3\sigma(I)$ $R_{int} = 0.058$ $\theta_{max} = 27.9^{\circ}$ $h = -14 \rightarrow 16$
$T_{min} = 0.767, T_{max} = 0.959$ 27028 measured reflections 3286 independent reflections <i>Refinement</i>	$k = -11 \rightarrow 10$ $l = -29 \rightarrow 29$
Refinement on F^2 $P[F^2 > 2\sigma(F^2)] = 0.022$	H atoms treated by a mixture

> 30(F)| $wR(F^2) = 0.081$ S=1.103286 reflections 173 parameters

ure of independent and constrained refinement $w=1/[\sigma^2({F_o}^2)]$ $(\Delta/\sigma)_{\rm max} = 0.001$ $\Delta \rho_{\rm max} = 0.29 \text{ e} \text{ \AA}^{-3}$ $\Delta \rho_{\rm min} = -0.23 \text{ e} \text{ Å}^{-3}$

Table 1	
Selected geometric parameters	(Å,

O1-N1	1.415 (2)	C1-C7	1.525 (3)
N1-C1	1.531 (2)	C2-C3	1.513 (2)
N1-C5	1.534 (2)	C3-C4	1.518 (2)
N2-C3	1.495 (2)	C4-C5	1.525 (2)
C1-C2	1.526 (2)	C5-C8	1.520 (2)
C1-C6	1.522 (3)	C5-C9	1.522 (2)
O1-N1-C1	108.2 (1)	N2-C3-C2	110.2 (1)
O1-N1-C5	109.6 (1)	N2-C3-C4	109.9 (1)
C1-N1-C5	119.3 (1)		
O1-N1-C1-C2	177.1 (1)	N2-C3-C4-C5	177.8 (1)
O1-N1-C5-C4	-176.2(1)	C1-N1-C5-C4	-50.8(2)
N1-C1-C2-C3	-53.5 (2)	C1-C2-C3-C4	60.8 (2)
N1-C5-C4-C3	53.4 (2)	C2-C1-N1-C5	51.0 (2)
N2-C3-C2-C1	-178.2 (1)	C2-C3-C4-C5	-61.1 (2)

°).

Table 2	
Hydrogen-bonding geometry (Å, °).	

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
N1-H10···Cl2	0.88 (2)	2.24 (2)	3.117 (6)	173 (2)
$O1-H11\cdots Cl2^i$	0.88 (2)	2.06 (2)	2.931 (5)	168 (2)
N2-H12···Cl1	0.98(2)	2.12 (2)	3.084 (2)	165 (2)
N2-H13···Cl1 ⁱⁱ	0.99 (2)	2.10 (2)	3.079 (2)	167 (2)
$N2-H14\cdots O2$	1.01 (2)	1.78 (2)	2.783 (2)	173 (2)
O2−H15···Cl1 ⁱⁱⁱ	0.90 (3)	2.24 (3)	3.135 (2)	170 (2)
O2−H16···Cl2 ⁱⁱⁱ	0.78 (3)	2.33 (3)	3.111 (5)	171 (3)
$C6-H6A\cdots O1$	0.98	2.42	2.812 (3)	104
$C7-H7A\cdots Cl2B$	0.98	2.70	3.531 (5)	142
C9−H9B···O1	0.98	2.45	2.863 (3)	105
$C9-H9B\cdotsO1^{iv}$	0.98	2.49	3.347 (2)	145

Symmetry codes: (i) 2 - x, -y, 1 - z; (ii) $\frac{3}{2} - x, \frac{1}{2} + y, z$; (iii) $2 - x, \frac{1}{2} + y, \frac{3}{2} - z$; (iv) -x, 1-y, 1-z

One chloride ion appears to be disordered, occupying two sites in a 60:40 ratio. All O-H and N-H H atoms were located in difference maps and refined isotropically, while all other H atoms were placed in calculated positions, with C-H distances of 0.98 Å and $U_{iso}(H) =$ $1.2U_{iso}(C)$ and were refined as riding.

Data collection: d*TREK (Rigaku/MSC, 2001); cell refinement: d*TREK; data reduction: d*TREK; program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: teXsan (Molecular Structure Corporation, 1992–1997); molecular graphics: teXsan; software used to prepare material for publication: teXsan.

References

Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115-119.

Blessing, R. H. (1995). Acta Cryst. A51, 33-38.

Hu, T. Q. (2003). J. Pulp Paper Sci. 29, 267-274.

Krishna, M. C., DeGraff, W., Hankovszhy, O. H., Sar, C. P., Kalai, T., Jeko, J.,

Russo, A., Mitchell, J. B. & Hideg, K. (1998). J. Med. Chem. 41, 3477-3492. Molecular Structure Corporation (1992-1997). teXsan. Version 1.8. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Rigaku/MSC (2001). d*TREK. Version 7.1I. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Sosnovsky, G. & Cai, Z.-W. (1995). J. Org. Chem. 60, 3414-3418.